

HOME SCHOOLING MATERIAL

PASS O' LEVEL

COMPUTER, MATH, CHEMISTRY

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COMPUTER STUDIES PAPER ONE QUESTIONS (OCOMPO06)

SECTION A

- Which component is responsible for converting the data 1. received from the user into computer understandable format?
 - Α Memory Unit
 - B. Arithmetic & Logic Unit
 - Input device
 - Output device D
- The only language which the computer understands is..... 2
 - assembly language Α
 - binary language В.
 - basic
- D. C language Which of the following is used to hold running program 3 instructions?
 - Flash disk Α.
 - В. Compact disc
 - С RAM
 - Hard disk D
- Which of the following programs would enable users to 4. perform specific tasks on a smartphone?
 - Android
 - Β. Disk cleaner C Antivirus
 - WhatsApp® D

5.

- What is the main difference between a mainframe and a super computer?
- Super computer is much larger than mainframe Α. computers
- В. Supercomputers handle complex computations while mainframe host a large amount of data
- C Super computers are much smaller than mainframe computers.
- D Supercomputers are focused to execute as many programs as possible while mainframe uses its power to execute few programs as fast as possible.
- The fifth generation of computers focuses much of the use 6. of

- Transistors A
- Β. Microprocessor
- С. Integrated Circuits
- D. Artificial intelligence Presentation, spreadsheet and photo-editing are examples of.
 - system software. A
- Β. operating system.
- application software.
- C. D.
- D. programming languages. After a picture has been taken with a camera on your 8 smartphone and processed appropriately, the actual print of the picture is considered.....
 - data.
 - А. В.
 - input С output
 - D. process
- 9 In. , an expert can recover and read deleted or damaged files from a criminal's computer.
 - Robotics A.
 - B. Simulation.
 - Forensics D Animation.
- 10. Choose the correct combination of the effects of a computer virus.
 - Disk Crash i.
 - Mother Board Crash ii
 - iii Corruption of program
 - iv Deletion of files
 - System unit crash v
 - (i), (ii), (iii) А
 - Β. (i), (iii), (iv)
 - C (ii), (iv), (v) D. (i), (iii), (v)
- In a spreadsheet program, =MAX (A2:A5) is an example of 11.
 - а..
 - value A Β. formula

- C. cell address D. function
- In order to send an e-mail to another class on the school's 12. local area network, a student must first know the class'.
- classroom teacher A.
 - В. e-mail password
 - computer type
 - D. e-mail address
- 13 When a file comprises of instructions that can be carried out by the computer, it is often called a(n)file.
 - A. Data
 - Β. information
 - executable C. D. memory
- 14. To move a picture within a word document, which procedure should a computer user follow?
 - Select, Paste, Print A.
 - Β. Select, Copy, Paste
 - C Select, Paste, Copy
 - D. Select, cut, Paste
- 15 Which of the following is designed to control the operations of a tablet?
 - Application Software А
 - System Software B Off shelf Software
 - C D. User
- Which advantage is offered to someone researching using 16. an online database instead of a printed set of similar
 - information?
 - Easier to access information Α.
 - Β. Better charts
 - C. Easier to understand tables D.
- More legible diagrams A computer user found a picture in a printed newspaper that 17 he wanted to use as a full-size cover for a report. Using the
 - computer, how might she do this? Print the report and glue the picture Α.
 - Turn to page II







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From page I

- Save the report and load the picture Load the report, add clip art, print B.
- C
- Scan the picture, resize, print D 18. What type of resource is most likely to be a shared common resource in a computer network? A. joystick

 - B. mouse
 - C. Keyboard
 - D. Printer
- 19. Which of the following controls the process of interaction between the user and the operating system?
 - User interface А.
 - В. Language translator
 - Platform D. Screen saver
- 20. A language translator that converts program statements lineby-line into machine language is called a(an).....
 - А. Interpreter
 - В. CPU C
 - Compiler Simulator D.

SECTION B

- 21. (a) Define a software licence? (b) Give two examples for each of the following type of software.
 - (i) Freeware
 - (ii) Shareware
 - (c) A Senior One student saw a presentation and
 - spreadsheet software on a computer. Explain the use of each of them.

 - (i) Presentation software (ii) Spreadsheet software.
- (a) What is the difference between data and information? 2.2 (b) Mention two devices in an information processing cycle that are at the specified levels of operation. (i) Input

- (ii) Output.
- (iii) Processing.(iv) Storage.

23

- (a) State any two examples of spreadsheet programs. (b) Explain three advantages of using a spreadsheet
- program.
- (c) Study the function provided =IF (A3>50, PASS, FAIL)
- State the category under which the above function (i) falls
- State whether the above function will run or not when executed. Give the reason for your answer. (ii) b) Give two functions which are in the same category as the
- one indicated above. 24 (a) (i) What is the difference between a web server and web browser (02 marks)
- (ii) Give two examples of web browsers (02 marks)
- (b) The screenshot below was obtained from a smartphone.



- What do the icons labelled A and B represent? (i) (ii) What is the major advantage of using connection A over connection B?
- (iii) With an example, define a protocol.25 (a) Define the following terms as used in storage.
 - Optical disc (i)
 - (ii) Disk cache
 - (b) The following specifications were obtained from a compact disc:
 - CDR-80, 700MB/80 Min (2x-56x CD-recordable) What does "80 Min" mean in these specifications.
 - (c) Sort the following storage media according to the

- categories given in the table. Flash disk, CD, Memory card, DVD, SD card, Blu-ray disc. Optical storage Solid state storage 26 (a) Explain the following network area coverages. Local Area Network. (i)
 - Wide area Network. (ii)
 - (iii) Metropolitan Area Network.(b) Explain the advantage of using each of the cables given
 - below.
 - (i) Fiber optic cable
 - (ii) Coaxial cable

SECTION C

- 27. (a) (i) What is a computer crime.
- (ii) Explain any four computer crimes. (b) Discuss five environmental threats/hazards to computers
- and information systems. 28. (a) Explain the following terms as used in programming. (i) An algorithm
 - (ii) Syntax

 - (b) Write an algorithm that prompts a user to enter two values, adds them and then displays the result.(c) Using a programming language of your choice, write a
 - program that can accept a value, obtains its square root and
- then displays the result. 29 (a) Explain the following terms as used in Information
 - technology.
 - troubleshooting (i)
 - (ii) control panel
 - (iii) cold booting (iv) uninstalling
 - (v) warm booting
 - (b) state any five components/ applets of the system control panel and their uses.

SOLUTIONS FOR MATHEMATICS PAPER 1 (OMTCO05)



ST MARY'S COLLEGE, KISUBI ST JOHN'S SS. NYABWINA - SHEEMA 4+18 = 22 M^{-1} = $\frac{1}{22}$ $\begin{array}{c}1\\6\end{array}$ 1 3 $\frac{1}{22}$ <u>3</u> 22 :.M⁻¹ = 11 Let the age of the Seventh boy be x4. 17 + 13 + 15 + 12 + 15 + 8 + x = 13 $\frac{80+x}{7} = 13$ 80 + x = (13x7)80 + x = 91x = 91-80x = 11:. The Seventh boy is 11 years. (x+2) and $(x-\frac{1}{3})$ are factors 5 (x+2) $(x-\frac{1}{3})=0$ $x^2 - \frac{1}{3} + 2x - \frac{2}{3}$

 $x^2 + \frac{5x}{3} - \frac{2}{3} = 0$

Multiply by 3 throughout $3x^2 + 5x - 2 = 0$ OR Sum of the roots $= -2 + \frac{1}{2}$ $=\frac{-5}{3}$ Product of the roots = $-2 \text{ x} \frac{1}{3}$ $= \underline{-2}$

 x^2 – (Sum of the roots)x + products of the roots = 0

$$x^2 - (\frac{5x}{3})x - \frac{2}{3} = 0$$

Multiplying by 3 gives $3x^2 + 5x - 2 = 0$ as the required quadratic equation

 $5x^4 - 80y^4$ $= 5(5x^4 - 16y^4)$ $= 5 ((x^2)^2 - ((2y)^2)^2)$ $= 5 [x^2)^2 - ((2y)^2] [x^2 + (2y)^2]$ $= 5 \left((x - 2y) (x + 2y) (x^{2} + 4y^{2}) \right)$ (15+-4)-4 1 0 5+0 11 5 = :. Therefore image of (5, -4) under the transformation matrix (3 1)is (11, 5) 1 0

7.



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12x + 2y = 0,400 (II)	
Solving (i) and (ii) simultar	neously
$2 \mid 5x+y=2,700$	(i)
1 12x+2y = 6,400	(ii)
10x + 2y = 5,400	(i)
12x+2y = 6,400	(ii)
Equation (ii) - Equation (i) g	ives
(12x-10x) + (2y-2y) = (6,400-5)	,400)
2x to = 1,000	
$\frac{2x}{2} = \frac{1000}{2}$	





15.



 $\begin{pmatrix} 28 & 0 \\ 0 & 28 \end{pmatrix}$

 $\frac{1}{28}$

b.(i) $P\theta =$

 $\theta P =$

 $: P\theta - \theta P =$

(ii)

17.

 $P\theta - \theta P =$

:. $(P+\theta)^2 =$

U

y

X=0

50

40

30

20

10

0

10 20 30

100,000 as the suitable value of I

5 x + 20y = 100

x + 4y = 20

Number of buses

 $10x + 20y \le 500$ $x + 2y \le 50$

 $x + y \leq 42$

 $y \leq 16$

 $\begin{pmatrix} -6 \\ 4 \end{pmatrix} \begin{pmatrix} x \\ y \end{pmatrix}$

 $\begin{pmatrix} x \\ y \end{pmatrix}$

 $\begin{pmatrix} 3 & -1 \\ 4 & 5 \end{pmatrix}$

 $\begin{pmatrix} x \\ y \end{pmatrix}$

 $x = \frac{1}{2}$ and $y = -\frac{1}{2}$

 $\begin{pmatrix} 3 & 1 \\ 0 & 2 \end{pmatrix} \qquad \begin{pmatrix} 3 & -1 \\ 4 & 5 \end{pmatrix} =$

 $\begin{bmatrix} 3 & -1 \\ 4 & 5 \end{bmatrix} + \begin{bmatrix} 3 & 1 \\ 0 & 2 \end{bmatrix}$

36

x > 0 and y > 0 (iv)

42 0

For (i) x + 2y = 50

25

For (ii) x + 2y = 42

0

42

For (iii) y = 16

 $\mathbf{X} + \mathbf{Y} = \mathbf{42}$

Number of lorries

5000x + 20,000y = 100,000 (i)

Income that is the objective function I=5000x + 20,000y. Using

Maximum point is (18, 16) that is, 18 lorries and 16 buses.

Maximum income = $(5000 \times 18) + (16 \times 20,000)$

= UGX 410.000

For (iv) x = 0 and y = 0

0

 $\begin{pmatrix} 4 & -6 \\ 2 & 4 \end{pmatrix} \quad \begin{pmatrix} x \\ y \end{pmatrix} = \frac{1}{28} \begin{pmatrix} 4 & 6 \\ -2 & 4 \end{pmatrix}$

 $\begin{pmatrix} 3 & 1 \\ 0 & 2 \end{pmatrix} =$

 $\begin{pmatrix} 9 & 1 \\ 12 & 14 \end{pmatrix} - \begin{pmatrix} 13 & 2 \\ 8 & 10 \end{pmatrix}$

 $= \begin{pmatrix} 5 \\ -1 \end{pmatrix}$

 $\frac{1}{28}$

 $\binom{1/2}{-1/2}$

 $\begin{pmatrix} 9 & 1 \\ 12 & 14 \end{pmatrix}$

13 2

(i)

(ii)

(iii)

Y = 16

40 50 60

X + 2Y = 50

(ii), is the search line

 $\binom{5}{-1}$

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MATHEMATICS PAPER (OMTCOO6) SECTION A

- Find the next two values in the sequences below 1 a) 1,1,2,3,5,8,13....., ...
- b) 11, 22, 110, 121, 202, 220,..., in base three *Y* varies inversely as the square of *x*. The difference 2 between the value of y when x = 6 and x = 10 is 16. Find the law connecting x and y
- The coordinates of A and B are (a 1) and (3, b) respectively. Given that (5, 2) is the midpoint of line 3. segment AB, find;

(a) column vector \overline{AM}

- (b) the length of vector \overline{AM} correct to 2 decimal places. Use logarithm tables to evaluate $\sqrt[3]{0.841 \times 32.7}$ correct to 3 significant figures.
- Express $\frac{1}{\sqrt{5}} + \frac{\sqrt{5}+2}{\sqrt{5}-2}$ in the form $a + b\sqrt{c}$. Hence state the 5 values of a, b and c.
- Given that $f(x) = \frac{x-1}{x^2 x 12}$. Determine the values of x for 6. which f(x) is undefined.
- 7. The ratio of new price to the old of a book is 5:4 Find the percentage increase in the price of the book a) If the old price of the book is UGX 3000, find the new b)
- price of the book In the figure below, \overline{PR} is parallel to \overline{AB} $\overline{AQ} = 4m$ $\overline{QR} = 21m$ $\overline{PR} = 14m$ and $\overline{AB} = 7m$.



Find ; a) angle x

b) length \overline{PQ} and \overline{QB}

- A map is drawn to a scale of 1: 200,000. Calculate the area of a field in km² that is represented by 2cm² on the map.
- A chord of length 16cm is 6cm away from the centre of a circle; calculate the circumference of the circle. $Use\pi = 3.14$

SECTION B

- 11. A survey was conducted on 30 employees to establish the transport means they preferred. It revealed that 7 preferred taxi (T) and Boda (B), but not Private cars (P), 5 preferred Taxi and private cars, but not Boda, 2 preferred Boda and private cars, but not Taxi, while 3 preferred neither of the transport means. Those who preferred private cars only were twice those who preferred all the 3 means of transport and a third of those who preferred Taxi only. 4 preferred Boda only. Represent this information on a Venn diagram.
 - b) From the Venn diagram, determine the how many people preferred;
 - (i) All the 3 means.
 - (ii) private cars.
 - (iii) taxi s.
 - c) Find the probability that a person chosen at random from the group used only one type of means.
- 12. (a) Given that $f(x) = \frac{x^2 1}{3}$, find;

(i) $f^{-1}(x)$

i)

ii)

(ii) The values of $f^{-1}(8)$.

(b) If $g(x) = x^2 - 2x - 1$ and h(x) = x + 3, find the value of x for which hg(x) - gh(x) = 2.

(a) without using a table or calculator, simplify the 13 following; $(16)^{\frac{-3}{4}}$, $(0,16)^{\frac{-1}{2}}$, $(5)^{-1}$

$$\left(\frac{1}{625}\right)^{-1} \times (0.16)^{2} \times \left(\frac{1}{2}\right)^{2}$$
$$\frac{1}{2} \log_{10} 16 - 2 \log_{10} \left(\frac{a}{5}\right) + \log_{10} a^{2}$$

Henry takes a train for two thirds of his journey, a bus b) for seven-eighth of the reminder and walks the rest of the journey. If his bus journey is 3km longer than what he walks. How long is the complete journey?

A conical hole is drilled in a solid wooden pyramid as shown in the figure below.



Given that $\overline{AD} = 6$ cm, $\overline{AB} = 10$ cm, depth of the cone $\overline{OF} =$ 4cm and $\overline{FE} = 2$ cm.

- Find (i) the maximum volume of water the cone contain. (ii) Volume of the remaining solid wood.
 - (iii) Surface area of the remaining solid wood.

5	Two cyclist A	Two cyclist A and B moved off from rest in the same										
	direction. The	direction. The speed of car A increases at a uniform rate										
	of 20kmh ⁻¹ w	of 20kmh ⁻¹ while the speed of B is as given in the table										
	below.			1			U					
	Time (hr)	0	1	2	3	4	5	6	7	8		
	Speed (kmh-1)	0	5	15	40	100	150	180	195	200		

- Using a suitable scale, draw on the same axes the a) velocity-time graph of cars A and B. b)
 - From your graph find the; (i)
- Time when the cars have equal speeds and the magnitude of the speed. Difference in speeds after 5 hours. Distance covered by car A in the 8 hours.
 - (ii)
- (iii)
- In the figure below, E is the mid-point of $\overline{AB}, \overline{OD}; \overline{DB} = 2:3$ and F is the point of intersection of \overline{OE} and \overline{AD}



Given that $\overline{OA} = \underline{a}$ and $\overline{OB} = \underline{b}$. Express the following vectors in terms of \underline{a} and \underline{b} . (i) \overline{AB} (ii) \overline{OE} (iii) \overline{AD}

Given further that $\overline{AF} = h\overline{AD}$ and $\overline{OF} = k\overline{OE}$. Find the two different expressions of \overline{OF} in terms of h, a, b. Hence find the scalars h and k.

Show that points O, F and E are collinear. 1. The table below shows a tax structure of some country.

Income sh	Tax rate %
1-50,000	Tax free
50,001-200,000	12
200,001-300,000	25
300,001-400,000	32
Above 400,000	40

Tax is levied on monthly income after the following allowances have been deducted.

Shs.45,000 for transport, Shs.64,000 for insurance,

Shs.630,000 annually for feeding, Shs.11,800 for water and electricity, *Shs.*40,000 for airtime and communication. Family allowances are also offered at the following rates; *Shs.*5,000 for each child below 10 years old and *Shs.*15,000

for each child aged between 10 and 19 years.

Mr. Madete earns a gross monthly salary of *Shs*.645,000 and has 5 children aged 7, 12, 18, 22 and 30 years. Family allowance is paid for only 3 children.

- Calculate; (a) His total monthly allowance.
 - (b) His income tax as a percentage of his gross monthly income.

(c) His net income.

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SOLUTIONS TO CHEMISTRY ISSUE 5 (OCHEM005)

- 1. Density of gases as compared to air. a.
 - Balloon X is displaced upwards because the gas therein is less dense than air. Balloon Y contains a gas b. which is denser than air so it is displaced down wards Gas X - upward delivery/downward displacement of c. air
 - Gas Y downward delivery/upward displacement of air.
 - d Used in advertisement balloons.
- 2. a) i) The two miscible liquid mixtures have different boiling points at which when they are heated, the liquids can be separated.
 - ii) Paper chromatography is based on the principle where molecules in mixture are applied onto the surface, the components have different speed along a stationary phase (stable phase) carried by a mobile phase. The more soluble ones will move more rapidly and further up the paper than the other less soluble component(s).
 - b) Air contains substance (gases and suspended matter) not chemically combined together and can be separated by physical means.
- a) i) A white ring is formed closer to the end with 3. hydrochloric acid (at point x)
 - ii) $NH_3(g) + HCl(g) \longrightarrow NH_4Cl(s)$

b) Gases or matter is made up of particles/molecules that are constantly moving. Lighter molecules move faster than heavier ones. The molecules of ammonia are lighter than the molecules in hydrogen chloride. The molecules in ammonia diffuse faster than those of hydrogen chloride, hence forming ammonium chloride (product) closer to the side with hydrogen chloride.

a) An isotope is a variant of a particular chemical element 4 which differs in neutron number and consequently in nucleon number.

h

Symbol	Name	Charge
nº	Neutron	0
p+	Proton	+1
e-	Electron	-1

b) Using the symbol T to represent tritium, give the formulae of (i) T

(ii) NaH

5. a) and b



c) When a catalyst is added, the activation energy is lowered. This lowers the energy required to form products hence the reaction is faster.

- 6. a) A redox reaction is a reaction in which one species is reduced and another is oxidised. b) Haematite contain iron (III) oxide, Fe_2O_5 . The oxygen is removed from the iron (III) oxide to form iron. This is a reduction reaction. The oxygen removed from haematite is added to carbon monoxide, forming carbon dioxide, this is an oxidation reaction. Since the process of extracting iron involves both reduction and oxidation reactions, it is a redox reaction c) $Fe_2O_3(s) + 3CO(g) \longrightarrow 2Fe(l) + 3CO_2(g)$
- 7. a) i) In the first set up, a pale green solution gradually



SEETA HIGH SCHOOL

BISHOP'S SENIOR SCHOOL, MUKONO

turned colourless

In the second setup, a colourless gas that burns with a blue flame and a pop-sound is evolved.

- ii) $Zn(s) + FeSO_4(aq) \longrightarrow ZnSO_4(aq) + Fe(s)$ b) Used to fill weather station balloons.
- Used to harden margarine
- Polypropene, a)

8.

- It can be used at temperatures beyond 100°C

It is insoluble in organic solvents like oil.b) It has low usability temperature. With hot food, polychloroethene bags melts and dissolves in organic food

- and this may cause ill health.
- c) i) High density polyethene
 - Food packaging Shopping bags
 - Dustbins crates
 - Water pipes

Low density polyethene

- Squeezable bottles
- Cling film milk carton lining Flexible water pipes cable jacketing

ii) Polyethene material is non-biodegradable. Poor disposal of polyethene leads to pollution of soil and blockage of drainage systems.

a) i) **Graphite** has a very high melting and boiling point because the strong covalent bonds that hold the carbon 9 atoms together in the layers require a lot of heat energy to break

ii) The layers in graphite can slide over each other because the forces between them are weak. This makes graphite slippery

- Making lead pencils.
 - Used as lubricant in electric devices.

10. a) the empirical formula of X

C	Н	0	
0.18	0.18	<u>0.96</u>	
1	1	16	
0.06	0.18	0.06	
<u>0.06</u> <u>0.18</u>		0.18	0.06
0.06 0.06		0.06	0.06
1	3	1	

The empirical formula is CH₃O

b) This is also called the hydration of ethene. Steam is reacted with ethene in presence of phosphoric acid as a catalyst at a temperature of about 300°C and pressure of 60 atmospheres

c) The orange solution of potassium dichromate turns green.

SECTION B

11. (a)(i) The reaction generates heat. Increasing or applying heat will lower the rate at which the products are formed. (ii) The reaction proceeds with increase in volume, since pressure and volume are inverse proportions. Increasing pressure will lower the rate of formation of products. (b) During the reaction, ammonia and oxygen are passed through a powdered catalyst. (i) Platinum

(iii) To increase the surface area upon which the reaction takes place

(ii) The catalyst lowers the activation energy required to form products. When applied, the reaction goes faster. c) At STP, 1 mole of a gas occupies 22.4dm³

- Moles of nitrogen monoxide:
- 22.4dm3 of a gas at STP occupy 1 mole of nitrogen

monoxide

(d)

a)

ii)

- 1dm³ of a gas at STP occupies <u>1</u> moles of nitrogen monoxide 720dm³ of a gas occupy $\frac{1}{22.4} \times \frac{720}{720}$ moles nitrogen monoxide
 - - = 32.143 moles of nitrogen monoxide

From equation, The mole ration of NO: HNO₃ is 1:1, hence the moles of nitric acid is 32.143 moles

Mass of nitric acid = moles of nitric acid × molecular formula of nitric acid 🛛 🗙

- $= 32.143 \times 63$
- = 2,025g of nitric acid

 $4NH_{3}(g) + 8O_{2}(g) \longrightarrow 4HNO_{3}(aq) + 4H_{2}O(g)$

- 12. a)i) Carbon burns in limited air to form carbon monoxide gas $2C (s) + O_2(g) \longrightarrow 2CO(g)$ $4NH3(g) + 502(g) \xrightarrow{Pt} 4NO(g) + 6H_2O(g)$

b) Manganese(IV) oxide catalyses the decomposition of aqueous hydrogen peroxide. In an experiment 50.0 g of maganese(IV) oxide. The total volume of oxygen formed was measured every 10 seconds. The results of the experiment are shown in the graph.



36 seconds

íi) At the start, the graph shows an exponential increase in the volume of oxygen formed with time, due to a high concentration of reactants. However, later, as the concentration of reactants reduce, the volume of oxygen formed lowers until there is no more increase in volume of oxygen formed when the reactants are used up.

- b) i) From the graph, it is 60 cm3 of oxygen gas
 - ii) 22400cm³ occupy 1 mole of oxygen at STP
 - $1 \text{ cm}^{3} \text{ occupies } \frac{1}{22400} \text{ moles of oxygen} \\ 60 \text{ cm}^{3} \text{ occupies } \frac{1}{22400} \times 60 \text{ moles of oxygen}$

= 0.002679 moles of oxygen

From equation: $2H_2O_2(aq) \longrightarrow 2H_2O(l) + O_2(g)$

1 mole of oxygen is produced by 2 moles of hydrogen peroxide 0.002679 moles of oxygen are produced by (2×0.002679) moles of hydrogen peroxide

= 0.005357 moles of hydrogen peroxide 50.0cm3 of solution contain 0.005357 moles of hydrogen peroxide

 1 cm^3 of solution contians $\underline{0.005357}$ moles of hydrogen peroxide

1000 cm³ of solution contains 0.005357×1000 moles of hydrogen peroxide

- = 0.107mol/dm⁻³ hydrogen peroxide
- 13. a) A colourless gas that turns lime water milky is given out. b) Calcium oxide is manufactured by the decomposition of calcium carbonate. i) $CaCO_3(s) \longrightarrow CaO(s) + CO_2(g)$

 - A drying agent in preparation of dry ammonia gas. ii)

c) i) Metal carbonate Identity Any group I metal carbonate e.g. Na₂CO₃ Mg____ Turn to page VI



PASS O'LENEL

Continued from page V

ii) Generally group 1 metal carbonates do not decompose when heated to give carbon dioxide, a basis for identifying Y. For other group II metals, reactivity increases down the group, so barium carbonate is more stable than magnesium carbonate since as you go down the group, the carbonate are more stable and therefore requires more time and heat to decompose them. d) i) $Ca(NO_3)_2(s) \longrightarrow CaO(s) + NO_2(g) + O_2(g)$

ii) Moles of calcium nitrate

 $\frac{\text{mass}}{\text{mass}} = \frac{0.10}{10} = 0.00061 \text{ moles of calcium nitrate}$ = Rfm 164

From equation, 1 mole of calcium nitrate liberates 1 mole of gas Moles of gas produced = 0.00061 moles (for either nitrogen dioxide or oxygen)



CHEMISTRY QUESTIONS (OCHEMO06)





a) i) Name the substance that was collected in tube P. ii) How is the substance named in a(i) formed? b)i) State what is observed in the first few minutes of the experiment.

- ii) Write an equation for the reaction.
- c) Give a suitable conclusion for the experiment in the set-up. Figure 5 shows an apparatus used to separate a mixture of water and hexene.



a) Name the apparatus.

2.

3

- b) State the principle by which the mixture of the two liquids is separated.
- c) i) Identify the liquids, R and S if the density of hexene is 0.66 g/cm3.
- ii) Give a reason for your answer in c(i) above. a) Distinguish between;
- i) Empirical and molecular formula of a compound ii) Atomic number and atomic massb) Element U has atomic number 12, while element V has
- atomic number 16. i) Write the electron configuration of the elements
- ii) Explain how the melting points of their oxides compare.
- The flow chart below represents some stages in the extraction of copper metal. Study it and answer the 4 questions that follow.

trend Creat 8,840

- a) Identify:
- i) The copper ore
- ii) Process B
- iii) Solid C

a)

b) i) Write an equation for the reaction that forms the slag. ii) State any one use of slag A solution contains 40.3g of substance XOH per litre.

250.0cm3 of this solution required 30.0cm3 of 0.3M sulphuric

- acid for complete neutralisation. Calculate the number of moles of:
 - Sulphuric acid i)
 - ii) XOH that reacted
- Determine the relative atomic mass of X. b) When burning magnesium ribbon was introduced into a 6. gas jar full of nitrogen,
 - a) i) State what was observed.
 - ii) Write an equation for the reaction
 - iii) Explain the observation in a(i) above. b) State one use of nitrogen.
- 7 The figure below shows a set-up used by a student to prepare dry chlorine gas in the laboratory.



- Identify two mistakes in the set-up, give a reason for each. b) i) Write an equation of reaction for preparation of chlorine gas
 - ii) State any two uses of chlorine gas.
- Explain why a dry litmus paper dipped in a jar c) containing chlorine gas will not be bleached, whereas a moist litmus paper is bleached. Study the flow chart below and answer the questions that
- 8. follow.

aqueous potassium salt	Barium Nitrate solution	Sodium M	Dilute Nitric (V) acid	Gas N
	ooranom		ciorci	

- Gas N forms a white suspension with aqueous calcium hydroxide. a) Name the anion present in the potassium salt. b) Write an ionic equation for the formation of solid M.
- c) Give one use of gas N. 30.0 cm³ of aqueous sodium hydroxide containing 8.0 g per litre of sodium hydroxide were completely neutralised by 0.294 g of a dibasic acid. Determine the relative formula
- mass of the dibasic acid. (Na = 23.0, O = 16.0, H 1.0) 10. One of the allotropes of sulphur is rhombic sulphur. a) Name the other allotrope of sulphur. b) Draw a diagram to show the shape of the allotrope named in (a) above.
 - c) i) State what is observed when concentrated sulphuric acid is reacted with sulphur.
 - ii) Write an equation for the reaction in c (i) above.

SECTION B:

- 11. a) Alkanes are said to be saturated hydrocarbons. i) What is meant by saturated hydrocarbons? ii) Draw the structure of the third member of the alkane homologous series and name it.
 - b) When the alkane, hexane, is heated to high
 - temperature, one of the products is ethene. i) Write the equation for the reaction.
 - ii) Name the process described in (b)

c) Study the flow chart below and answer the questions that follow



- i) Identify A.
- ii) Write an equation that leads to formation of D iii) Draw the structure of D
- iv) Give a reason why D pollutes the environment.
- v) Write an equation for the formation of F.

- d) Name one reagent which can be used to distinguish butene from butanol and state what is observed in each case. The diagram below was used to prepare hydrogen
- chloride gas which was passed over heated iron powder.



- i) Give a pair of reagents that will produce hydrogen a) chloride gas in flask A.
- ii) Name the substance in flask B
- b) i) State what is observed in the combustion tube.ii) Write an equation for the reaction in the combustion tube.
- c) i) Identify the gas that burns at the jet.ii) Write an equation for the reaction in c (i) above.
- d) i) Give reasons why excess hydrogen chloride gas is dissolved using the funnel arrangement.
- ii) Describe how hydrogen chloride gas can be tested in the laboratory.

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